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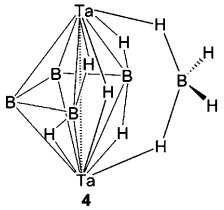
In the Full Paper by S. Ghosh et al., mistakes have been found in the interpretation of proton resonances for compound **4**. Previously, it was deduced that the resonance at $\delta = -10.9$ ppm corresponded to four protons, which was incorrect. The corrections are as follows: On page 13487, in paragraph 2, as the temperature is lowered the resonance at $\delta = -10.9$ ppm collapses into four resonances. The two resonances appearing at $\delta = -10.7$ and -11.4 ppm should be assigned to two Ta–H–B protons, the resonance at $\delta = 4.15$ ppm is assigned to two B5–H_t (t = terminal) protons, and the signal at $\delta = -15.9$ ppm is assigned to Ta–H_t. These results illustrate that Ta1 and Ta2 become inequivalent at the lowest experimental temperatures due to the tantalum hydride. Therefore, the correct chemical formula of compound **4** in the abstract; on page 13486 and in the legend of Figure 5; on page 13488 and in entry 6 of Table 2; and in the Table of Contents should be $[(Cp^*Ta)_2B_4H_9(\mu\text{-}BH_4)]$. In the abstract, on page 13483, and on page 13486, $[(Cp^*Ta)_2B_4H_8]$ should be replaced with $[(Cp^*Ta)_2B_4H_9]$. In the Experimental Section on page 13489, the following corrections should be made to the synthesis of **4**: ^{1}H NMR: $\delta = -10.91$ (br, 5H); elemental analysis calcd (%) for $^{12}C_{20}{}^{1}H_{43}{}^{11}B_5{}^{180}Ta_2$: C 34.34, H 6.20. Additionally, the schematic diagram of **4** in Figure 7 on page 13488 is shown correctly here. The authors sincerely apologize for these oversights and thank Prof. R. B. King for bringing this error to our attention.



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